



Investigation of the Effects of Gasoline-Bioethanol Blends on Engine Performance and Exhaust Emissions in a Spark Ignition Engine

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Abstract

In this study; 95 octane unleaded gasoline and bioethanol were tested as fuel in a spark-ignition engine by blending bioethanol with gasoline in certain proportions (2%, 4%, 6%, 8%, 10%). Tests were carried out at five different engine loads for a constant engine speed of 2500 rpm. The variations in engine performance parameters (engine torque, engine power, brake thermal efficiency, brake specific fuel consumption, brake specific energy consumption) and exhaust emissions (exhaust gas temperature, HC, CO, CO₂ and O₂) were investigated on the basis of test fuel and engine load. According to the test results, with the increase of bioethanol ratio in the blend, brake specific fuel consumption values increased by 9.71% and brake thermal efficiency values decreased by approximately 4.97% compared to gasoline. There was decrease up to 35.56% and up to 23.77% in HC and CO emissions respectively, and an increase by 6.01% in CO₂ emissions with bioethanol addition.

Keywords: Gasoline, bioethanol, exhaust emission, engine performance

1. INTRODUCTION

Energy is one of the fundamental and driving needs of a countries economic and social development. In this respect, “Energy Security” is a vital element of economic and national security. It is an indispensable input for almost all the processes necessary to sustain our social life and used in sub-sectors such as industry, transportation, housing and commercial. Today, 86% of the energy consumption is met from fossil sources such as oil, natural gas and coal in the world. Oil, as the main energy source of the transportation sector, has the largest share in the world’s primary energy consumption. Turkey is one of the significant energy consumers with its emerging economy. Oil ranks first with the rate of 31% in primary energy demand which is equal to 136.2 MTEP in 2016 and it is followed by natural gas and coal. When the distribution of Turkey’s primary energy demand was examined, it was observed as follows: 25% of it in industry, 24% of it in housing and service industry, 23% of it in electricity generation, 20% of it in transportation industry [1].

The fact that most of the energy demand is met from petroleum and its derivatives, however, the depletion of oil reserves, directs humanity to long-lasting, no consumable resources. Considering the environmental impacts caused

by petroleum fuels such as carbon dioxide and greenhouse gas emissions and acid rain associated with them, global warming and climate changes, the importance of alternative energy increases. The use of alternative energy sources is also inevitable for Turkey which meet the oil demand with importing 89% of it [1, 2].

The petroleum crisis at 1970s led many countries to searching of alternative fuels, thus the use of ethanol as a fuel has been brought to the agenda, and studies on fuel ethanol have increased rapidly day by day. In the early days, corn was used as the raw material of ethanol production because of its easy production and its ability to be converted to a high rate to alcohol [3, 4]. Today, renewable products such as sugar cane, sugar beet and cassava are the most important raw materials of ethanol. Bioethanol has same molecule with ethanol as chemically. The difference of bioethanol from ethanol is that it produces from biological feedstocks via fermentation methods. Therefore, it called bioethanol. The ethanol produces from petrochemical feedstocks such as ethylene, calcium carbide, coal, natural gas via catalytic hydration [5-7].

Since bioethanol has similar fuel properties with gasoline, it can be used both directly and mixed with gasoline in spark plug ignition engines. In addition, the higher octane number of ethanol allows spark ignition engines to run at higher

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compression ratios. Many European countries with primarily the world's largest bioethanol producers USA and Brazil, also use bioethanol as fuel [8].

The use of bioethanol as fuel causes a decrease in engine performance due to its low calorific value [9]. However, there are also many studies showing that exhaust emissions of spark ignition engines reduce by increasing the combustion quality thanks to the oxygen content of bioethanol [10, 11]. It is even stated that, thanks to its oxygen content, it increases the volumetric efficiency and thus causes an increase in the thermal efficiency values from time to time [12, 13].

Altun, Öztop [14] investigated the effects of ethanol and methanol addition (5, 10% by volume) to unleaded gasoline on performance and exhaust emissions of a spark-ignition engine. They performed their experiments at different engine speeds varied from 1000 to 4000 rpm for full throttle position. It was observed decrease in brake thermal efficiency, CO and HC emissions, while brake specific fuel consumption increased with alcohol addition. It stated that researchers achieved best results in the case of the engine being fueled blend contain 10% methanol or ethanol at current engine design. Balki, Sayin [15] examined the use of alcohols (ethanol and methanol) on performance, emission and combustion characteristics of a single-cylinder engine by comparing unleaded gasoline. They performed the experiments at variable engine speeds for full open throttle. As a result, they stated that the use of alcohol fuels increases engine torque, instantaneous specific fuel consumption, thermal efficiency and combustion efficiency. In the case of engine being operated with alcohols, the cylinder gas pressure and heat release rate had already occurred before. It also reported that CO₂ emission increased while HC, CO and NOx emissions decreased. Najafi, Ghobadian [16] have tested the engine with E5, E7.5, E10, E12.5 and E15 gasoline-ethanol blends to investigate the performance and emission behavior of a four-cylinder, four-stroke spark ignition engine. They stated that usage of ethanol in blend increased brake torque, and brake power and improved the exhaust emissions but it decreased brake specific fuel consumption. The optimum values are obtained when engine operated at 3000 rpm engine speed with E10. Özsezen [17] used a single-cylinder spark-ignition engine with a compression ratio of 8/1 in their experimental study. Within the context of study, tests were performed at 1000, 1500, 2000 and 2500 rpm in a spark-ignition engine fueled with gasoline-ethanol (E5 and E10) blends to examine engine performance and exhaust emissions. The results of study indicated that CO and unburned HC emissions were reduced but CO₂ and NOx emissions were increased when the test engine was run with gasoline-ethanol blends. In addition, it is stated that the air-fuel ratio increased with increasing ratio of ethanol in blended fuels. Deng, Chen [18] carried out their tests in a single-cylinder, variable compression ratio spark-ignition engine operated with unleaded gasoline, pure ethanol and their blends (10% and 20%). Test conditions are as follows:

five different compression ratios by of 4/1, 5.5 /1, 7/1, 8.5/1, 10/1 and full open throttle. It was found that brake average effective pressure, brake thermal efficiency and brake specific fuel consumption were higher when the fuel is gasoline-ethanol blend. They also stated that it was obtained better results for gasoline-ethanol blends compared with pure gasoline with regard to exhaust emissions. They also added the results that NOx emissions became more unstable as the compression ratio changed during the tests.

Bioethanol is a renewable alcohol with high energy value that can be produced with local sources. It is used as alternative fuel in spark plug ignition engines due to the high octane number. In this study, the effects of up to 10% bioethanol addition to gasoline on performance and emission parameters were analyzed step by step. The engine torque, power, brake specific fuel consumption, brake specific energy consumption and brake thermal efficiency values were investigated as engine performance parameters, and CO, HC, CO₂, O₂ and exhaust gas temperature values were investigated as exhaust emission characteristics. The obtained results were compared with gasoline, and have been presented as graphically.

2. MATERIAL AND METHOD

The test setup consisting of universal drive and brake unit, display panel, fuel tank, measuring tube for fuel consumption, temperature measuring device, pressure measuring device, speed measuring device, emission measuring device and a gasoline engine is shown in Figure 1. Technical specifications of the single cylinder, four stroke and air-cooled internal combustion engine are given in Table 1.

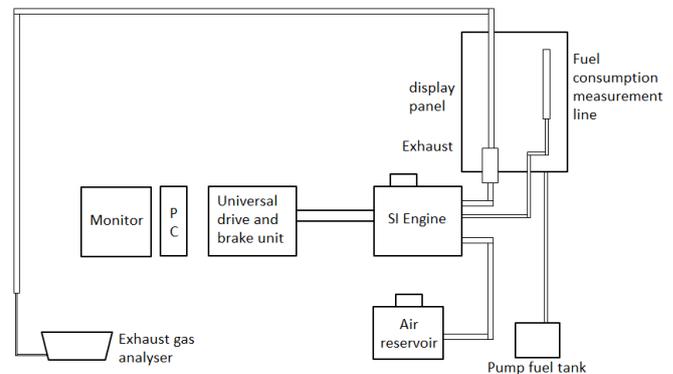


Figure 1. Experimental setup

Table 1. Specification of test engine

Engine Brand	GUNT CT152
Engine type	Four stroke, with carburettor
Cylinder number	1
Bore x Stroke, (mm x mm)	65.1 x 44.4
Cooling system	Air cooled
Compression ratio	7:1
Maximum engine power	1.2 kW
Maximum engine torque	4.5 Nm
Ignition advance	25° BTDC

Measurement ranges and accuracies of Mobydic 5000 portable gas analyzer used to measure exhaust emissions are given in Table 2.

Table 2. Specifications of the Mobydic 5000 portable gas analyzer

Measurement module	Ranges	Accuracies
CO (% vol)	0-10	0.01
CO ₂ (% vol)	0-20	0.01
HC (ppm vol)	0-2000	1
NO _x (% vol)	0-5000	1
Lambda	0-5	0.001

In the experiments, 95 octane unleaded gasoline purchased from one of the gas stations and bioethanol obtained supplied from Konya Sugar Factory were used to form fuel blends, contents of which are as follows. In Table 3, some physical properties of test fuels are given.

E0: 100% unleaded gasoline

E2: 98% unleaded gasoline - 2% bioethanol

E4: 96% unleaded gasoline - 4% bioethanol

E6: 94% unleaded gasoline - 6% bioethanol

E8: 92% unleaded gasoline - 8% bioethanol

E10: 90% unleaded gasoline - 10% bioethanol

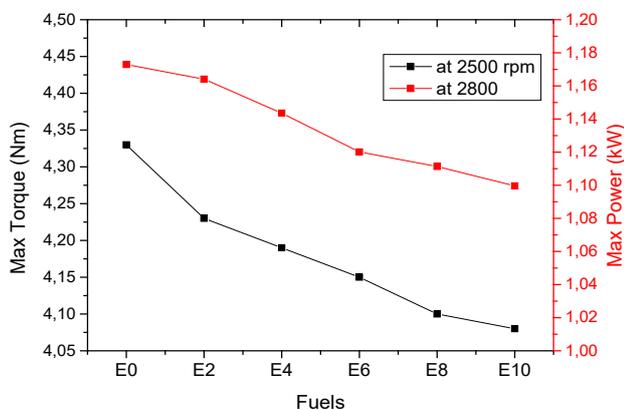
Table 3. Some physical properties of test fuels

Test fuels	Density (15°C-g/cm ³)	Lower heating value (MJ/kg)	Kinematic viscosity (40°C-mm ² /s)	Water content (ppm)
Bioethanol (E100)	0.78820	26.694	1.2	1093.4
Gasoline (E0)	0.72926	42.582	0.566081	-
E2	0.73198	42.131	0.576054	-
E4	0.73354	41.899	0.587451	-
E6	0.73557	41.732	0.597899	-
E8	0.73594	41.235	0.607872	-
E10	0.73809	40.901	0.617845	-

3. RESULTS

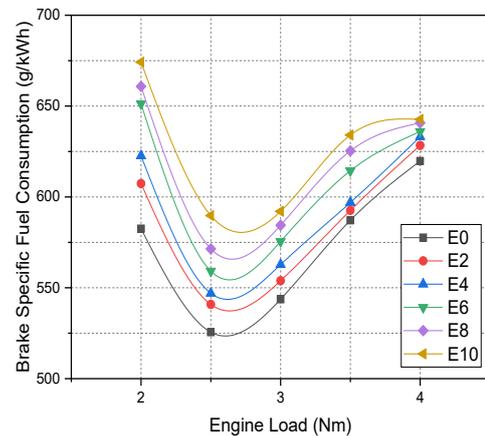
3.1. Engine Performance Parameters

Figure 2 shows changes of max torque and max power values for test fuels. The lower of calorific value of bioethanol caused a decrease both torque and power values. This decrement increased with increment of bioethanol ratio in blends. The highest decrement values have obtained by 5.77% for torque and by 6.25 for power with E10 fuel.

**Figure 2.** Maximum torque and maximum power ratings of test fuels

Brake specific fuel consumption (BSFC) is the amount of fuel, used by the engine to obtain brake power. BSFC values obtained during the tests performed at five engine loads

and constant engine speed of 2500 rpm, maximum torque speed, are shown in Figure 3. As a result of the experiments, the lowest BSFC values were obtained at 2.5 Nm engine load for all test fuels. When the data obtained for all engine loads were examined, the lowest BSFC value was obtained as 525.593 g/kWh with gasoline. With the increase in bioethanol ratios in the blends, this value increased to 589.765 g/kWh with E10 fuel. The highest BSFC values for gasoline, E2 and E4 fuels are obtained at 4 Nm engine load, while the BSFC values at 4 Nm were lower than those at 2 Nm for other test fuels. This can be explained by the fact that the ideal operating range of the engine is around 2.5 Nm - 3 Nm. Load average-BSFC values are higher approximately 2.27%, 3.65%, 6.26%, 7.91% and 9.70% for E2, E4, E6, E8 and E10 respectively than that of E0. The bioethanol has lower calorific value (about 37.31%) than gasoline. This caused the lower thermal value of the E10 fuel to drop to approximately 4%. Therefore, BSFC values for bioethanol added fuels are higher than gasoline. In the literature, there are studies showing that addition of bioethanol increases BSFC unless any modification is made in the engine for similar reasons[19, 20].

**Figure 3.** Variation of BSFC with engine load

Brake specific energy consumption (BSEC) is the amount of energy consumed by engine to generate unit power. The variation of BSEC with engine load and bioethanol content are shown in Figure 4. As a result of the experiments, the lowest BSEC values for all test fuels were obtained at 2.5 Nm engine load. When the data obtained for all engine loads were examined, the lowest BSEC value was obtained with gasoline as 22.381 MJ/kWh, this value reached up to 24.122 MJ/kWh for E10 fuel for the same engine load value by increasing with bioethanol rate. For gasoline, E2 and E4 fuels, the highest BSEC were obtained at 4 Nm engine load, and for other fuels BSEC are lower at 4 Nm than that of 2 Nm. This can be explained by the fact that the energy per unit power is lower in the range of 2.5 Nm to 3 Nm engine loads. For average engine loads, the increase in BSEC are approximately 1.18%, 1.99%, 4.14%, 4.49% and 5.37% for E2, E4, E6, E8 and E10 respectively with respect to gasoline. Both the increase in fuel consumption and decrease in calorific value with bioethanol addition have led to an increase in the BSEC. Zhuang and Hong [21] also reported in their studies that due to the low calorific value of bioethanol, the amount of heat ener-

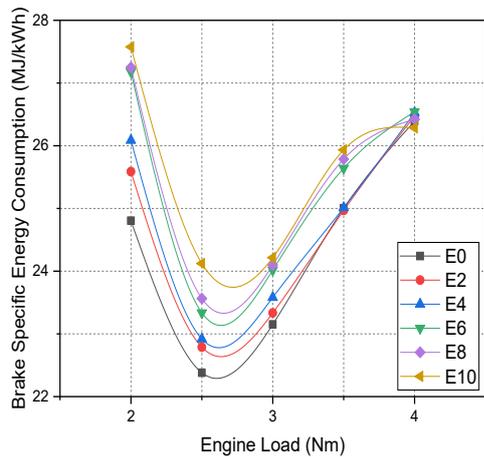


Figure 4. Variation of BSEC with engine load

gy released in the cylinder decreased and fuel consumption increased with bioethanol usage, and therefore BSEC values increased.

Figure 5 shows the variation in brake thermal efficiency (BTE), which is an expression of conversion ratio of fuel energy to brake power. As a result of the experiments, the highest BTE values were obtained with gasoline and it was observed that the BTE values decreased with increasing bioethanol ratio in the blend. For all test fuels, maximum BTE values were obtained at 2.5 Nm engine load i.e. 16.085% for gasoline and 14.924% for E10 fuel. After 2.5 Nm engine load, BTE values have started to decrease with the increase in friction losses and the amount of fuel consumed. With bioethanol usage, the decrease in BTE are approximately 1.16%, 1.92%, 3.90%, 4.21% and 4.97% for E2, E4, E6, E8 and E10 respectively. As a result of the decrease in the thermal value of the addition of bioethanol, it requires more fuel consumption per unit time in order to obtain an equal amount of brake power under the same operating conditions of engine. Due to the fact that the amount of fuel taken into the cylinders in one cycle is almost the same for all test fuels and lower heating values of the blended fuels, the amount of energy generated in the cylinder during combustion decreases, thus reduce the engine output power. Since this decreasing engine output power value is obtained with the same amount of fuel, the BTE values of the engine are also decreased in the calculations. In the literature, some researchers have reported that due to the high evaporation temperature of bioethanol in the combustion chamber cannot be achieved in the homogeneity of the mixture so the use of bioethanol decreased BTE values [22, 23].

3.2. Exhaust Emission Parameters

The exhaust gas temperature (EGT) is an important parameter in the interpretation of exhaust emission parameters and the quality of combustion within the cylinder. The variation in the EGT depending on the engine load are shown in Figure 6. By adding bioethanol to gasoline, EGT values increased up to 473.2 °C. It is the result of increasing combustion efficiency affected positively by oxygen content of bioethanol. However, Topgül [24] stated that addition of ethanol in ra-

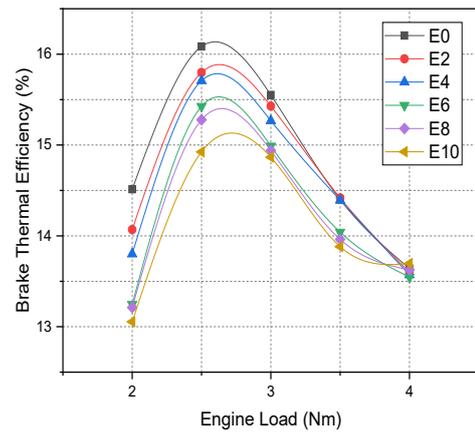


Figure 5. Variation of BTE with engine load

tios above 10% ethanol undermines the homogeneity of the mixture by the effect of high evaporation heat of it. It is also reported that increased oxygen content of the blend reduced the amount of hydrocarbon and increased combustion rate decreased the end-combustion temperature. According to the results of the experiment, with the use of bioethanol, the EGT values of E2, E4, E6, E8 and E10 fuels increased by 0.66%, 1.83%, 3.02%, 3.36% and 3.70% on average. With the increase in engine load, more fuel enters to the cylinder at the same speed, but the time required to burn it cannot be provided. Therefore, the fuel does not burn completely and the EGT decreases.

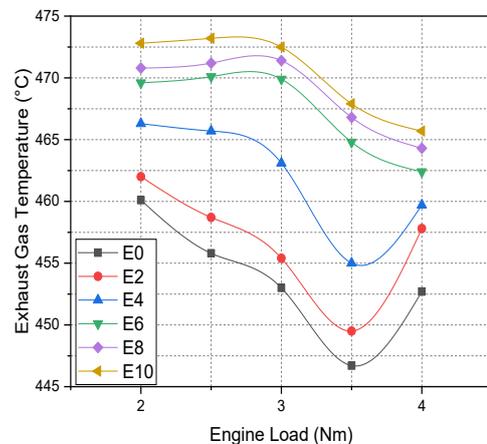


Figure 6. Variation of EGT with engine load

With the combustion of hydrocarbon-based engine fuels, H_2O_{gaz} produce as usual. However, if sufficient oxygen is not taken into the cylinder, the H atom cannot find enough O atoms for this reaction and generates HC emissions. Bioethanol allows the H atoms in the fuel to break off from the C atoms during combustion and react with oxygen to burn. Thus, the combustion approaches the completion. As shown in Figure 7, HC emission values decreased to 172 ppm with the use of bioethanol. In addition, the increase in EGT is another factor in reducing HC emissions. Increased HC emissions due to engine load are the result of reduced EGT as engine load increases. While the lowest HC emission values were obtained with E10 fuel, as the bioethanol content in the mixture increased, HC emission values decreased by 5.33%, 11.16%, 22.70%, 27.70% and 35.56% with

E2, E4, E6, E8 and E10 fuels on average. Similar results have been reported in the literature showing that bioethanol reduces HC emissions [25, 26].

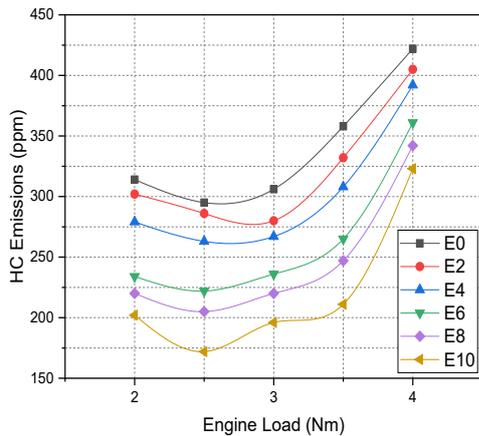


Figure 7. Variation of HC emission with engine load

Low oxygen intake to the cylinders prevents the complete combustion of the fuel and releases CO gas from the exhaust instead of CO₂, usual combustion product. The increase in engine load requires more fuel to burn in the same duration. The greater amount of fuel in the combustion chamber cannot burn faster, CO emissions increase with increasing engine load. Figure 8 shows variation of CO emission with engine load. Bioethanol supply O₂ to react CO during combustion, resulting in a reduction in CO emissions. For example, CO emissions were reduced by up to 2.29% with E10 fuel compared to gasoline. Load averaged-CO emission values decreased by 7.86%, 11.71%, 22.72%, 23.77% and 26.04% on average with E2, E4, E6, E8 and E10 fuels. It has been reported by other researchers that bioethanol reduces CO emissions as it causes the fuel burn completely [10, 27, 28].

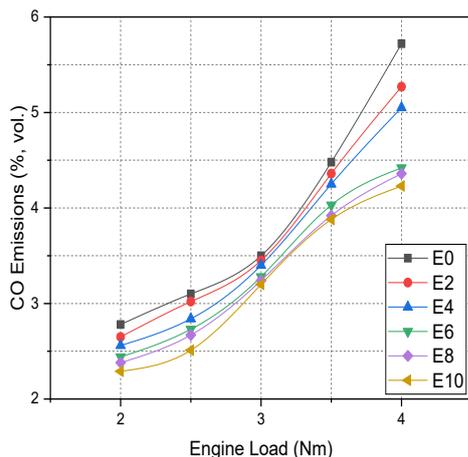


Figure 8. Variation of CO emission with engine load

Figure 9 shows the variation of CO₂ emission depending on engine load. Bioethanol has small amount of C atoms compared to gasoline. It improves the combustion efficiency, resulting as a reduction in CO₂ emissions by 6.01%. However, the lowest CO₂ emission values were obtained at 3.5 Nm engine load, where combustion efficiency was close to ideal, and CO₂ emissions were increased due to the decrease in

the amount of fuel taken into the cylinder at lower engine loads and the homogeneity of the mixture due to the high evaporation temperature of bioethanol. In addition, at higher engine loads, the time for the fuel to burn was shortened, thus causing the fuel to not burn completely, causing CO emissions. These results are similar to previous studies in the literature [15, 29-31].

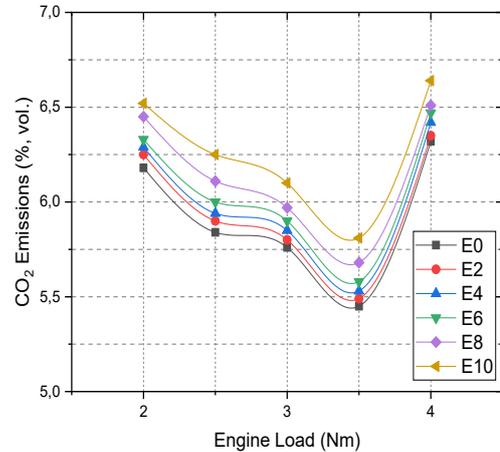


Figure 9. Variation of CO₂ emission with engine load

Variations of O₂ emission at different engine load values are given in Figure 10. The O₂ concentration depends on the oxygen/fuel ratio in the cylinders. Bioethanol contains oxygen in its chemical structure and increases the oxygen concentration of the mixture fuels made with gasoline. In this experimental study, O₂ emission values of blended fuels were higher than gasoline. The highest O₂ emission for all test fuels are found for E10 fuel as 12.03%. The variation in engine load had no significant effect on the amount of O₂ in the exhaust gas. Compared to gasoline, O₂ emission values of E2, E4, E6, E8 and E10 fuels are higher as by 1.96%, 2.48%, 11.43%, 12.03% and 22.39% respectively.

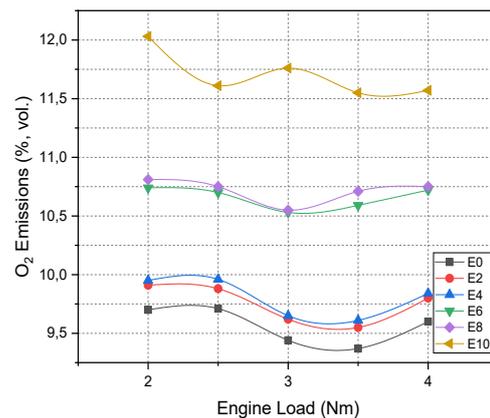


Figure 10. Variation of O₂ emission with engine load

4. CONCLUSION

In this study, engine performance and exhaust emission of a single cylinder spark ignition engine used gasoline-bioethanol blends were examined and the results were presented as follows.

- Although bioethanol produced from sugar beet has

similar fuel characteristics with gasoline, the lower heating value of it stands out as the most important parameter adversely affecting engine performance.

- Due to the fact that it has lower thermal value than gasoline and higher combustion rate due to oxygen content, bioethanol has a negative effect on performance parameters by decreasing in the effective pressure affecting the piston surface. The increase in bioethanol ratio in the blends resulted in an increase in the BSFC and BSEC values up to 9.70% and 5.37% on average, while the BTE values decreased by of 4.97%, respectively.
- Blending gasoline with bioethanol up to 10% improves the combustion efficiency, thus an increase in exhaust temperature values observed up to 3.7% compared to gasoline. It indicates that the fuel can burn more completely, thus it also reduce HC emissions.
- Bioethanol causes cleaner combustion and reduces harmful exhaust gases. Oxygen in bioethanol content reacted with H and C atoms in hydrocarbon based fuels during combustion, so it causing less HC and CO emissions by of 35.56% and 26.04%, respectively. In addition, it has a reduction in CO₂ emissions up to 6.01% as it has a lower amount of C atoms than gasoline.

In conclusion, although the addition of bioethanol to gasoline affects the engine performance parameters negatively, it is seen that this negative effect is at most 9.70%. Compared to it with improvement in exhaust emission parameters by of 35.56%, bioethanol is a convenient fuel. Furthermore, the fact that addition bioethanol by of 10% to fuel can be used without any change in engine operating parameters indicates that legal limit (4%) for addition bioethanol to gasoline can be easily increased in our country.

5. ACKNOWLEDGMENT

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