

Investigation of *In Vivo* Effects of Carbon Tetrachloride (CCl₄) and Quercetin on Some Metabolic Enzyme Activities in Rat Erythrocyte

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Keywords

Carbon tetrachloride, Quercetin, Glucose 6phosphate dehydrogenase, 6phosphogluconat e dehydrogenase, Glutathione reductase, Glutatyon Stransferaz. **Abstract:** In the study; the purpose was to investigate the *in vivo* impact of carbon tetrachloride (CCl₄) and quercetin (Qu) on activities of important metabolic enzymes such as Glucose 6-phosphate dehydrogenase (G6PD), 6-phosphogluconate dehydrogenase (6PGD), glutathione reductase (GR) and glutathione S-transferase (GST) in rat erythrocytes. At the experimental stage, rats were divided into 4 groups. 1.Group (Control): Pure olive oil at a dose determined according to their body weight (1mL/kg) was given to the rats in this group, 2.Group (CCl4: 1.0 mL/kg (ip)(1:1), 3.Group (Ku: 25) mg/kg (ip), 4.Group (CCl4(1.0 ml/kg (ip)+ Ku (25 mg/kg (ip) was injected. The study was continued for 3 days. The results revealed that the activities of; G6PD (p<0.01), 6PGD (p<0.01), GR(p<0.001) and GST (p>0.05) enzyme activities were decreased in the CCl₄ group compared to the control group. It was determined that enzyme activities were higher in CCl₄+Qu applied groups compared to CCl₄ group. The application of Qu caused an increase in the enzyme activity value. This can be accepted as an indication that the inhibition caused by CCl₄ has disappeared. Consequently; It is thought that Qu may be effective in preventing oxidative damage due to CCl₄ administration.

Sıçan Eritrositlerinde Karbon Tetraklorürün(CCl4) ve Kuersetinin Bazı Metabolik Enzim Aktiviteleri Üzerine *İn Vivo* Etkisinin İncelenmesi

Anahtar Kelimeler	Öz: Çalışmada; Sıçan eritrositlerinde karbon tetraklorür (CCl ₄) ve Kuersetinin(Ku)'in, Glukoz
Karbon	6-fosfat dehidrogenaz (G6PD), 6-fosfoglukonat dehidrogenaz (6PGD), glutatyon redüktaz (GR)
tetraklorür,	ve glutatyon S-transferaz (GST) gibi önemli metabolik enzim aktiviteleri üzerine in vivo
Kuersetin,	etkilerinin incelenmesi hedeflendi. Deneysel aşamada sıçanlar 4 gruba ayrıldı. 1.Grup(Kontrol):
Glukoz 6-Fosfat	Bu gruptaki ratlara vücut ağırlıklarına göre belirlenen dozda(1 mL/kg)saf zeytin yağı,
Dehidrogenaz, 6-	2.Grup(CCl ₄ : 1.0 mL/kg (i.p.)(1:1), 3.Grup(Ku: 25 mg/kg (i.p),4.Grup(CCl ₄ (1.0 ml/kg (i.p.)+
fosfoglukonat	Ku (25 mg/kg (i.p.) enjekte edildi.Çalışma 3 gün sürdürüldü. Sonuçlar değerlendirildiğinde;
dehidrogenaz,	G6PD (p<0,01), 6PGD (p<0,01) GR (p<0,001) ve GST (p>0,05) enzim aktiviteleri kontrol
Glutatyon	grubuna oranla CCl ₄ grubunda azalmıştır. CCl ₄ +Ku uygulanan gruplarda ise enzim
redüktaz,	aktivitelerinin CCl4 grubuna oranla yüksek olduğu belirlenmiştir. Ku uygulanması, enzim
Glutatyon S-	aktivite değerinin yükselmesine neden olmuştur. Buda CCI4'ün neden olduğu inhibisyonun
transferaz.	ortadan kalktığının göstergesi olarak kabul edilebilir. Sonuç olarak, CCl4 uygulamasına bağlı
	olarak ortaya çıkan oksidatif hasarı önlemede Ku'nın etkili olabileceği düşünülmektedir.

1. INTRODUCTION

Carbon tetrachloride (CCl₄) is a colorless, nonflammable, quickly evaporating, fragrant dense liquid, and is widely used in the production of petroleum products, varnishes, lacquers, resin solvents and organic compounds [1]. It is frequently used in dry cleaning, firefighting, grain disinfection and insect control [2]. Biologically inactive CCl₄ is transformed into reactive toxic metabolites for their activation and takes part in target cells [3,4,5]. CCl₄ exerts its toxic effect with the formation of a free radical, the trichloromethyl radical. The peroxyl radical, which is formed as a result of the combination of this radical with oxygen, is a strong lipid peroxidation initiator that plays a role as the primary mechanism in the formation of cell damage by disrupting the cell membrane structure [3,6,7,8]. Malondialdehyde (MDA), which is formed as the end product of lipid peroxidation, is frequently used in the determination of oxidative damage [9]. In CCl₄ poisoning; analogous to oxidative damage as an experimental model [10] (Figure 1).



Figure 1. Molecular structure of carbon tetrachloride (CCl₄) [11].

Quercetin (Qu) is one of the most widely studied bioflavonoids in the flavonols class [12] and is one of the most common flavonoids found and many dietary sources [13]. Qu, occurs mainly in the leaves and other parts of the plant in the form of glycosides or aglycones formed by the attachment of one or more sugar groups to phenolic groups by glycosidic bonds. Ou has powerful anti-oxidative and cytoprotective impact on oxidantinduced endothelial cell apoptosis due to its chemical structure [12]. Qu, which is included in various herbal teas today; It is used in biochemistry, food chemistry, paint chemistry, medical chemistry, paint industry and cosmetics[14,15]. Qu, prevents oxidative damage and cell death by prohibit lipid peroxidation and scavenging free oxygen radicals [16,17]. Qu, exerts its possible biological properties through its antioxidant activity [18] (Figure 2).



Figure 2. The structure of quercetin [19].

G6PD and 6PGD are known as metabolic enzymes that catalyze the first and third steps of the pentose phosphate pathway. As an agent that protects cells against oxidative stress, NADPH is known as an important molecule for biochemical processes such as reducing biosynthesis reactions [20,21]. NADPH is the most important molecule that reduces the disulfide form of glutathione,

as well as being considered as a cellular reducer, which is significant for nitric oxide biosynthesis, detoxification of xenobiotics and pharmaceuticals [22-24]. GR preserve the cell counter the harmful effects of these molecules by reducing oxidized molecules thanks to the -SH groups in its construction. In addition, membrane proteins such as GSH, the reduced form of GR, hemoglobin and spectrin, and the preserve of thiol groups of various enzymes, DNA and protein synthesis, detoxification of xenobiotics and some metabolic end products, and transport of amino acids [25,26]. The GR enzyme catalyzes the regeneration reaction of glutathione in the presence of NADPH, thereby maintaining the intracellular redox balance [27]. The binding reactions of GSH to endogenous and exogenous hydrophobic electrophiles are carried out by GST isoenzymes. GSTs can noncatalytically bind many exogenous and endogenous compounds such as fatty acids, hormones, flavonoids, bilirubin and xenobiotics [28,29]. The purpose of the presented study was to investigate the in vivo effects of CCl₄ and Qu treatments on some metabolic enzymes G6PD, 6PGD, GR and GST activities in rat erythrocytes.

2.MATERIALS AND METHODS

2.1.Chemicals

The chemicals used in the study were purchased from Sigma and Merck.

2.2. Experimental Design

Twenty-eight Wistar albino male rats weighing 200-300 g were used. They were fed ad-libitum with standard laboratory chow and water at a constant temperature of $20 \pm 2^{\circ}$ C and twelve (12 hours) light-dark cycle (light; 07:00-19:00, dark 19:00-07:00). The rats were randomly divided into 4 groups of 7 in each group. The study was approved by the Bingol University Experimental Animal Ethics Committee. (BUHADEK: 4.10.2018-2018/08-08/01).

1. Group (Control): Pure olive oil i.p. given through in a dose determined according to the body weight of the rats (1 mL/kg) was given to the rats in this group [30].

2.Group (CCl₄): The rats in this group were mixed with 1 mL/kg of CCl4 in olive oil at a ratio of 1:1 and administered i.p. was applied as. [31].

3.Group (Qu): The rats in this group were administered i.p. at a dose of 25 mg/kg by using the stock solution prepared in dimethyl sulfoxide (DMSO) [32].

4.Group (CCl₄ + Qu): The rats in this group were administered 1 mL/kg CCl4 (1:1 in olive oil) + 25 mg/kg Qu (using the stock solution prepared in DMSO) i.p. At the end of the 3rd day, blood samples were taken from the rats under anesthesia.

2.3. Preparation of Hemolysate

Blood samples were taken into EDTA tubes. Fresh blood samples were taken into EDTA tubes. Afterwards, plasma and leukocytes were eliminated by centrifugation for 15 minutes (2500 xg). Red cells packed with KCl solution (0.16 M) were washed three times. Blood

samples were centrifuged at (2500 xg) and supernatants were discarded. Erythrocytes were hemolyzed using 5-fold distilled water. Samples were centrifuged at $+4^{\circ}$ C (10000 xg) for 30 minutes to eliminated cell membranes and insoluble molecules. The supernatant was saved for analysis [33].

2.4. Determination of Enzyme Activities

G6PD and 6PGD enzyme activity was measured spectrophotometrically according to the Beutler method at 340 nm. This method is based on the principle that NADPHs in the reaction medium absorb light at 340 nm. In reactions catalyzed by both G6PD and 6PGD, enzymes react with NADP+ as a co-substrate and convert to NADPH. This increase is followed spectrophotometrically, the amount of absorption per minute is found and converted into enzyme units [34-36]. The method proposed by Carlberg and Mannervik was used for the measurement of GR enzyme activity. This method depends on the reduction of NADPH in the reaction catalyzed by the GR enzyme. This decrease was monitored by spectrophotometry at 340 nm to determine enzyme activity [37] and the activity of the GST enzyme was determined according to the Habig method. This method is based on the conversion of 1-chloro-2,4dinitrobenzene (CDNB) to S-glutathione dinitrobenzene (DNB-SG). The product of this reaction, DNB-SG, shows maximum absorbance at 340 nm [38].

3. STATISTICAL ANALYSIS

Statistical evaluation was made using the SPSS 20 program. Data were expressed as mean \pm SD. The Kruskal Wallis test followed by the Mann Whitney U test was used to define the diverse among the groups. The diverse among the groups was remerkable important when p< 0.05.

4. RESULTS AND DISCUSSION

According to the results of the current study; G6PD enzyme activity was statistically lower in the CCl₄ group compared to the control group (p<0.01). Qu applied group; G6PD enzyme activity was found close to the control group.In the CCl₄+Qu group, the G6PD enzyme activity was discover to be higher than the CCl₄-treated group (Figure 3)



Figure 3. The in vivo effect of CCl₄ and Qu on rat erythrocyte G6PD enzyme activity (a,b differs between groups with different letters) (p<0.01).

When the 6PGD enzyme activity was assessed, a statistically significant reduction was observed in the CCl₄ treated group compared to the control group (p<0.01). The 6PGD enzyme activity was found to be higher in the Qu-treated group than in the CCl₄ group. Enzyme activity rised in CCl₄+Qu group compared to CCl₄ group. These results show that Qu reduces the inhibitory effects of CCl₄ on 6PGD enzyme activity(Figure 4).



Figure 4. In vivo effect of CCl₄ and Qu on rat erythrocyte 6PGD enzyme activity (a,b, difference between groups with different letters) (p<0,01).

When the GR enzyme activity is evaluated; it was observed that the enzyme activity was reduced in the CCl₄ applied group compared to the control group(p<0,001). In the CCl₄ +Qu applied group, it was identified that the GR enzyme activity was increased compared to the CCl₄ group (Figure 5).



Figure 5. In vivo effect of CCl_4 and Qu on rat erythrocyte GR enzyme activity (a,b,c there is difference between groups with different letters) (p<0.001).

When the GST enzyme activity was assessed; it was determined that there was a decrease in the CCl₄ group compared to the control group, and a partial increase was observed in the CCl₄+Qu group compared to the CCl₄ group. No statistical differences were discovered between the groups (p>0,05) (Figure 6).



Figure 6. In vivo effect of CCl₄ and Qu on rat erythrocyte GST enzyme activity (p>0,05).

When the results of this study are evaluated in general; G6PD, 6PGD and GR enzyme activities are inhibited by CCl₄. As a result of prevention of G6PD and 6PGD enzyme activities, cells cannot produce enough NADPH in the pentose phosphate pathway to detoxify endogenous and exogenous oxidized molecules. At the same time, inhibition of GR and GST enzyme activities may cause disruption in intracellular GSH metabolism, thus rendering cells vulnerable to oxidative stress. In the study; G6PD enzyme activity was decreased in CCl₄ group compared to control group (p<0.01). Enzyme activity in the Qu group was discover to be close to the control group. In the CCl₄+Qu group, the enzyme activity was discover to be higher than in the CCl₄ group. This suggests that quercetin may have a decreasing effect on the inhibition of CCl4 in G6PD enzyme activity. When 6PGD enzyme activity is evaluated; The decrease in enzyme activity (p<0.01) in the group treated with CCl₄ compared to the control group can be considered as an indicator of inhibition of the 6PGD enzyme by CCl₄ application. The rise in the enzyme activity value in the CCl₄+Qu applied group can be considered as an indication that the inhibition effect is partially eliminated. While it was observed that GR enzyme activity reduction statistically important (p<0.001) in the carbon tetrachloride group compared to the control group, it was concluded that the enzyme activity increased partially in the CCl₄ +Qu administered group. When the GST enzyme activity was assessment, a partial reduction was determined in the CCl₄ group compared to the control group. It was seen that the increases or decreases between the groups were not statistically significant. In the literature review, no study was found that examined the effects of CCl₄ and Qu on metabolic enzymes in rat erythrocytes. In a different study, it was determined that carbon tetrachloride (CCl₄) intoxication caused a decrease in enzyme activities such as NADPH-cytochrome c reductase, NADH-cytochrome reductase, coumarin hydroxylase, 7-ethoxy coumarin-Odeethylase, UDP-glucuronyl transferase and glutathione-S-transferase [39]. In a different study by Sheweita et al., it was determined that CCl₄ application caused inhibition of GST enzyme activity [40]. In a study by Elbe et al., the antioxidative effects of Qu in CCl₄-caused cardiac detriment were histologically and biochemically investigated. While SOD activity were remarkably declined in CCl₄ administered group, Qu administration caused considerable rise in SOD activity. In CCl₄ toxicity, it was concluded that the use of Qu could provide beneficial effects owing to its antioxidative properties [41]. In a different study investigating the antioxidant defense system in tissue damage caused by CCl₄ and ethyl alcohol in goose liver and kidneys, which are among the important antioxidant and metabolic enzymes due to damage to the liver and kidneys, glutation peroxidase(GSH-Px), catalase (CAT), glucose-6-phosphate dehydrogenase (G6PD) activities were decreased [42]. In a different study by Jeon et al., it was reported that carbon tetra chloride administration caused a decrease in the activities of important metabolic enzymes glutathione S-transferase and glutathione reductase in the rat liver [43]. In a different study researchers evaluated teh effects of CCl₄

and Qu treatment. They demonstrated that Qu isolated from *Delonix elata* L. plant has significant prophylactic effects on liver function markers AST, ALT, ALP, serum bilirubin and total protein, as well as antioxidant enzymes SOD, CAT, GPx and GST [44]. Literature also stated that after CCl₄ exposure, antioxidant enzyme activities such as SOD, CAT and GPx decreased significantly in the rat kidney, CCl₄ exposure caused oxidative stress by inhibiting antioxidant enzyme activities [45]. There are many other studies revealing various drugs and substances inhibit activities of metabolic enzymes [46-48].

5. CONCLUSION

When the results of this study are evaluated in general; G6PD, 6PGD and GR enzyme activities are inhibited by CCl4. In the CCl₄+Qu applied groups, the enzyme activities were higher than the CCl₄ group. While the GST enzyme activity decreased in the CCl4administered group, it partially increased in the Qutreated group. These results indicated that Qu reduces the inhibition impact of CCl₄ on G6PD, 6PGD and GR enzyme activities. Accordingly, it is thought that Qu may be effective in preventing oxidative damage due to CCl₄ administration. At the same time, there is a need for further studies in which different Quercetin dosages will be used.

CONFLICTS OF INTEREST

There is no conflict between the authors.

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128