

Anadolu Briyoloji http://dergipark.org.tr/tr/pub/anatolianbryology Dergisi **Research** Article DOI: 10.26672/anatolianbryology.1705249 e-ISSN:2458-8474 Online



Anatolian Bryology

Detection of some phenolic content of Grimmia ovalis (Hedw.) Lindb. and investigation of its antimicrobial activity with molecular docking technique

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| Received: 23 May 2025 | Revised: 17 June 2025 | Accepted: 18 June 2025 |
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#### Abstract

This study aims to investigate the phenolic compounds and pharmacological potential of Grimmia ovalis (Hedw.) Lindb. to understand their bioactive potential and to pioneer future research. It also aims to be one of the first studies on G. ovalis's in silico bioactivity. HPLC analysis revealed the detection of some phenolic compounds in G. ovalis extract. The study revealed the in silico antimicrobial potential of some phenolic compounds. We calculated that compounds of 2,5-dihydroxybenzoic acid and caffeic acid would inhibit Staphylococcus aureus UDP-N-acetylenolpyruvylglucosamine reductase (MurB) at micromolar levels. We concluded that they interacted with residues crucial for antimicrobial activity on the mentioned macromolecule.

Keywords: Bryophyte, Grimmiaceae, Antimicrobial Activity, Phenolic Compounds, Molecular Docking

# Grimmia ovalis (Hedw.) Lindb. Türünün Bazı Fenolik Bileşiklerinin Belirlenmesi ve Moleküler Yerleştirme Tekniği ile Antimikrobiyal Aktivitesinin Araştırılması

Öz

Bu calışma, Grimmia ovalis (Hedw.) Lindb.'in fenolik bileşenlerini ve farmakolojik potansiyelini araştırmayı, biyoaktif potansiyellerini anlamayı ve gelecekteki araştırmalara öncü olmayı amaçlamaktadır. Bununla beraber in silico biyoaktivite acısından G. ovalis üzerine yapılan ilk calısmalardan biri olmayı hedeflemektedir. HPLC analizi, G. ovalis özütünün bazı fenolik bileşiklerinin tespitini ortaya koymuştur. Bazı fenolik bileşiklerin in silico antimikrobiyal potansiyeli ortaya çıkarılmıştır. 2,5-Dihidroksibenzoik asit ve Kafeik asit bileşiklerinin, Staphylococcus aureus UDP-N-asetilenolpiruvil glukozamin redüktaz (MurB) üzerinde mikromolar düzeyde inhibisyon yapacağı hesaplanmıştır. Bahsi geçen makromolekül üzerinde antimikrobiyal aktivite için önemli olan kalıntılar ile etkileşime girdiği sonucuna varılmıştır.

Anahtar kelimeler: Briyofit, Antimikrobiyal Aktivite, Fenolik Bileşikler, Moleküler Doklama.

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To cite this article: Cambay Z. Uslu H. Göktas B. Özdemir Bayçınar K. Tükenmez Emre Ü. Alatas M. 2025. Detection of some phenolic content of Grimmia ovalis (Hedw.) Lindb. and investigation of its antimicrobial activity with molecular docking technique. Anatolian Bryology. 11:1, 52-58.

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#### 1. Introduction

Bryophytes, or non-vascular plants, comprise a substantial and varied category that includes mosses, liverworts, and hornworts, and inhabit numerous ecological niches in terrestrial environments. The genus *Grimmia*, part of the family Grimmiaceae, is a significant ecological group of mosses, predominantly located on rocks and arid surfaces in temperate and hilly areas. However, despite their prevalence in ecosystems, studies on the phytochemical contents and pharmacological potential of *Grimmia* species are quite limited (Delgadillo Moya, 2015; Glime, 2007; Hardman, 2007).

Bryophytes have recently garnered significant scientific interest due to their unique secondary metabolites and resilience to extreme environmental conditions. The displayed features indicate a promising source of bioactive compounds with therapeutic applications (Asakawa and Ludwiczuk, 2018; Rios and Recio, 2005). Research focused on the *Grimmia* genus remains limited.

Initial research indicates that certain *Grimmia* species, including *Grimmia pilifera*, exhibit significant antibacterial and antioxidant properties, potentially attributable to the presence of flavonoids, diterpenes, and other phenolic chemicals (Barwant and Tripathi, 2023; Elibol et al., 2011; Peters et al., 2018). In addition, some species of this genus are considered to have the potential to be a source in the formulation of antibacterial drugs.

The aim of this study is to obtain information on the *in silico* antimicrobial activity of phenolic compounds obtained from the extract of *G. ovalis*. Furthermore, to our knowledge, this study is intended to be one of the first studies on the phytochemical and bioactivity of *G. ovalis*. This will help us understand how it can be used in pharmacology and open the door to future uses in natural product drug discovery.

# 2. Materials and Methods

# 2.1. Plant material and extraction process

The powdered plant was macerated with 70% ethanol for 3 days by rotating on a rotavapor for 1h. After filtering, it was evaporated to dryness on a rotavapor. The yield of the extract was determined as 6.15%.

The research materials were collected from Dolin Surroundings (Niğde), located in the C13 (Figure 1). Square in Turkey according to the Henderson (Henderson, 1961) grid system. It is a moss belonging to the Bryophyta division, Bryopsida class, Grimmiales order, Grimmiaceae family. It is one of the 33 taxa of the *Grimmia* genus in our country (Kürschner and Frey, 2011). *G. ovalis* is a taxon that generally spreads cosmopolitanly on rocks and loves semi-neutral, arid and open environments (Table 1). The bryophyte specimen (*G. ovalis*) used in the study is preserved in the bryophyte collection of ALATAŞ (Tunceli) with the Herbarium number ALT 4620.

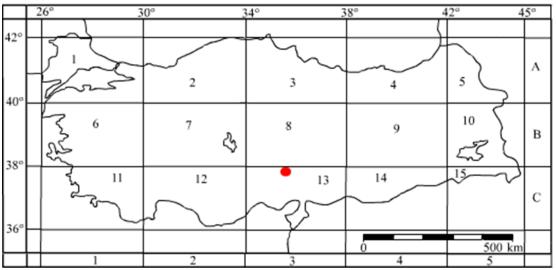


Figure 1. According to the Turkish grid system, the locality where the Grimmia sample was taken

| Table 1. Locality information       |              |            |                          |  |
|-------------------------------------|--------------|------------|--------------------------|--|
| Locality                            | Altitude (m) | Date       | GPS Record               |  |
| Dolin Surroundings, Niğde (Red Dot) | 1690         | 14.09.2024 | 34°43'6.37"E, 37°53'40"N |  |

#### 2.2. Phenolic content

HPLC grade methanol was purchased from J.T. Baker, Acetonitrile was purchased from Isolab and all other chemicals were purchased from Merck. 8.0 mg (399.75 mg/ml) of the above ground parts of G. ovalis was taken and closed in an airtight bottle. Alliance e2695 HPLC and C18 column (250 x 4.56mm.  $5\mu$ ) was used to determine the phenolic compounds of the aboveground parts of G. ovalis. Phosphoric acid (pH 2) and Methanol:Acetonitrile (90:10) were used as mobile phases. The flow rate was set as 0.8 ml/min. The wavelength was determined as 280 nm and photodiode array (PDA) detector was used. The injection volume was 20 µL and the temperature was 30°C. The results were obtained with the Empower 3 program (Pirisi, Cabras, Cao, Migliorini, and Muggelli, 2000; Veneziani et al., 2018).

### 2.3. Molecular docking

The compounds detected in the phenolic components of G. ovalis were plotted to get **SMILES** from PubChem (https://pubchem.ncbi.nlm.nih.gov). The two substances found in the highest amounts (2,5-Dihydroxybenzoic acid and Caffeic acid) in the extract were identified and the molecular docking process was applied to these substances. Energy minimization of two components were performed with the ChemOffice software. To evaluate 1HSKantimicrobial activity, molecular docking studies were performed using the standard procedure to determine the binding modes and docking scores of the two compounds detected in G. ovalis extract two phenolic components at the active sites of the 1HSK (Benson et al., 2001). The macromolecule crystal structures was retrieved from the Protein Data Bank server (https://www.rcsb.org/, accessed 20 May 2025) and optimized with Schrödinger Maestro (Maestro, 2024). Molecular docking was performed with both Autodock and Vina softwares (Eberhardt, Santos-Martins, Tillack, and Forli, 2021; Huey, Morris, Olson, and Goodsell, 2007). Since we had previously worked with this macromolecule for 1HSK, FAD was re-docked into the target site of the macromolecule to validate the docking program and the RMSD value was found to be appropriate (<2) (Unver, Uslu, Gurhan, and Goktas, 2024).

# 3. Results and Discussion 3.1. Phenolic content

According to the results of the applied HPLC analysis, 6 phenolic components were determined *G. ovalis* extract (Table 2). Two of these compounds were detected at very high

concentrations compared to the others. 2,5-Dihydroxybenzoic acid was found at 1.219 ppm and Caffeic acid was found at 0.492 ppm and. Other compounds obtained were detected in the range of 0.251 ppm to 0.030 ppm.

### 3.2. Molecular docking

The active binding sites of the macromolecule (Pdb ID: 1HSK) have been previously determined in the protein data bank (Benson et al., 2001). Docking studies were performed to see the interaction modes the two phenolic compounds (2,5of Dihydroxybenzoic acid and Caffeic acid) detected at the highest concentration in the G. ovalis extract with the active site of the macromolecule. Binding types and associated aminoacids were generated in detail by Maestro Software (Table 3-4, Figure 2-5). Some residues previously identified as important for the interaction Staphylococcus aureus UDP-Nacetylenolpyruvylglucosamine reductase (MurB) were described in detail in previous study (Benson et al., 2001; Unver et al., 2024). In the receptor, waters around the previously identified active site FAD (10Å) were left and all other water molecules were removed. The regular spacing of the grid boxes was determined to be 0.375 Å centered on FAD (40x40x40 Å<sup>3</sup>). The interaction modes with 1HSK for 2,5-Dihydroxybenzoic acid and Caffeic acid were visualized in 2D and 3D with the Maestro program.

In conclusion, this study was carried out to reveal the in silico antimicrobial potential of phenolic components of G. ovalis extract. HPLC analysis revealed the detection of some phenolic compounds. Molecular docking study revealed dock scores and interacting residues. Molecular docking scores were good with 1HSK, which was selected for the antimicrobial activity study. In addition, some of the interactions observed and reported to be important between 1HSK and the cofactor (FAD) in previous studies were also observed in the two phenolic compounds we obtained and used in molecular docking studies. These interactions were hydrogen bonding with ASN80 and GLY146 for 2,5-Dihydroxbenzoic acid, and hydrogen bonding with TYR77, ASN80 and SER143 for Caffeic acid.

Consequently, this study, in which some phenolic contents of G. *ovalis* were determined and its antimicrobial activity was investigated by molecular docking technique, will constitute a resource for such studies on bryophytes in the future.

| No | Compound                  | Chemical Structure | ppm   | <b>R</b> <sup>2</sup> |
|----|---------------------------|--------------------|-------|-----------------------|
| 1  | 2,5-Dihydroxybenzoic acid | ОН О<br>ОН<br>ОН   | 1.219 | 0.999979              |
| 2  | 3,4-Dihydroxybenzoic acid | НО ОН              | 0.251 | 0.997677              |
| 3  | Caffeic acid              | НО ОН              | 0.492 | 0.999713              |
| 4  | Vanillic acid             | ОННО               | 0.161 | 0.999671              |
| 5  | Syringic acid             | О ОН НО ОН         | 0.071 | 0.983057              |
| 6  | <i>p</i> -Qumaric acid    | ОН                 | 0.03  | 0.999736              |

Table 2. Chemical compositions of the phenolic components of G. ovalis extract.

 Table 3. Molecular docking scores and estimated inhibition constants of Chemical compositions of phenolic components of G. ovalis and 1HSK.

|                            | <b>Autodock Results</b>                 |                          | Vina Results             |  |
|----------------------------|---|--------------------------|--------------------------|--|
| Compounds                  | Estimated<br>Inhibition<br>Constant, Ki | Best<br>Docking<br>Score | Best<br>Docking<br>Score |  |
| 2,5- Dihydroxybenzoic acid | 24.63 µM                                | -6.29                    | -6.8                     |  |
| Caffeic acid               | 5.94 µM                                 | -7.13                    | -7.5                     |  |

µM: micromolar, Docking Score: Estimated Free Energy of Binding (kcal/mol)

Table 4. Interacting residues and interaction types of two phenolic components of G. ovalis with 1HSK.

| ~ .                        | Autodock Results     |                   |  |
|----------------------------|----------------------|-------------------|--|
| Compounds                  | Interacting Residues | Interaction Types |  |
| 2,5- Dihydroxybenzoic acid | ASN80                | Hydrogen Bond     |  |
|                            | GLY146               | Hydrogen Bond     |  |
|                            | SER143               | Hydrogen Bond     |  |
|                            | TYR77                | Hydrogen Bond     |  |
| Caffeic acid               | ASN80                | Hydrogen Bond     |  |
|                            | SER143               | Hydrogen Bond     |  |

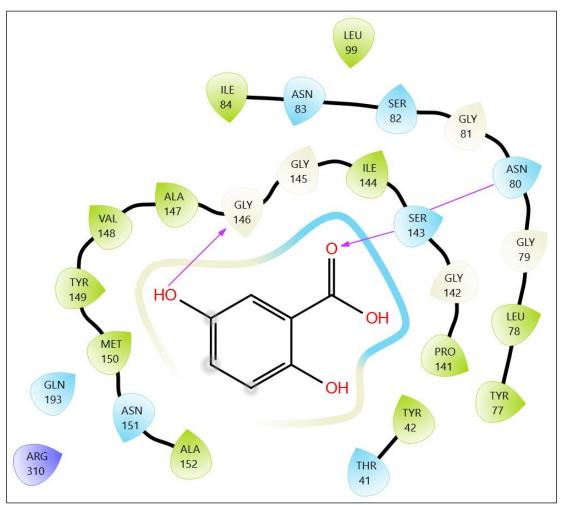


Figure 2. 2D interaction diagram with 1HSK for 2,5-Dihydroxbenzoic acid (Purple arrow: H-Bonds).

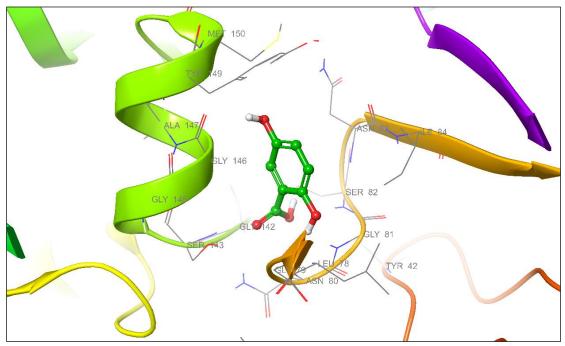


Figure 3. 3D interaction diagram with 1HSK for 2,5-Dihydroxbenzoic acid.

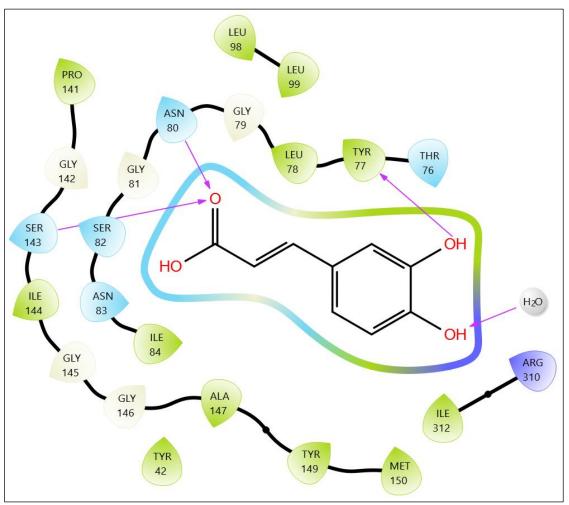


Figure 4. 2D interaction diagram with 1HSK for Caffeic acid (Purple arrow: H-Bonds).

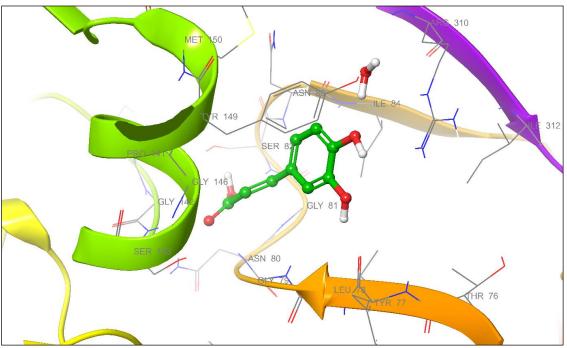


Figure 5. 3D interaction diagram with 1HSK for Caffeic acid.

#### Declaration

Author contributions: Concept, ZÇ; Conception and design, ZÇ, MA, HU; Supervision consultancy, MA; Resources, BG, KÖB, ÜTE; Materials, MA, ZÇ, ÜTE; Data collection and/or processing, ZÇ, KÖB, BG; Analysis and/or interpretation, ZÇ, HU, BG; Literature search, ZÇ, KÖB, MA; Writing stage, ZÇ; Critical review, MA, ZÇ.

**Conflict of interest:** The authors have no competing interest to declare regarding the content of this article.

**Funding:** The authors declare that no funding, grant or other support was received during the preparation of this article.

**Ethical approval:** This research did not involve human or animal subjects and therefore does not require ethical approval.

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